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phosphorus de . MOLES AND MASS Name Determine  
the number of moles in each of the quantities below. 1.  
25 g of Naa 2. 125 g of 3. 100. g of 4 74 g of KCl z LB  
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quantities below. 1) 25 g of NaCl 2) 125 g ... 6) 2.5 moles of NaCl 7) 0.50 moles H<sub>2</sub>SO<sub>4</sub> 8) 1.70 moles of KMnO<sub>4</sub> 9) 0.25 moles of KCl 10) 3.2 moles of CuSO<sub>4</sub>•5H<sub>2</sub>O

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0.21 mole Directions: determine the number of grams in each of the quantities below. 6) 2.8 moles of NaCl ANSWER: 160 g 7) 0.75 moles H<sub>2</sub>SO<sub>4</sub> ANSWER: 74 g 8) 2.75 moles of KMnO<sub>4</sub> ANSWER: 435. g 9) 0.55 moles of KCl ANSWER: 41. g 10) 2.3 moles of CuSO<sub>4</sub>•5H<sub>2</sub>O ANSWER: 570 g Chemistry IF8766 Page 50 Instructional Fair, Inc. Moles and Mass - Ms. Agostine's Chemistry Page Answer Key Chemistry If8766 Moles And Mass Free PDF eBook Download: Answer Key Chemistry If8766 Moles And Mass Download or Read Online eBook answer key chemistry if8766 moles and mass in PDF Format From The Best User Guide Database Moles and Mass. Directions: Determine the number of moles in each of the quantities below. 1) 52



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